# PATENT SPECIFICATION

NO DRAWINGS

Inventors: JOHN DAVID KENDALL, GEORGE FRANK DUFFIN and JACQUES VOLTZ





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#### COMPLETE SPECIFICATION

## Diazacarbocyanines

We, J. R. GEIGY S.A., a Swiss Company of Basle, Switzerland, do hereby declare the invention for which we pray that a patent may be granted to us, and the method by which it is to be performed, to be particularly described in and by the following statement:—

This invention relates to new diazacarbocyanine dyes.

In our co-pending Application No. 20414/
58 (Serial No. 885,520) there is described inter alia, a new class of diazacarbocyanine bases of the general formula I:—

$$R_1$$
  $R_2$   $R_2$   $R_2$   $R_3$   $R_4$   $R_4$   $R_4$   $R_5$   $R_6$   $R_6$ 

wherein  $R_1$  is an alkyl or aralkyl group,  $R_2$  is an alkyl, aryl or aralkyl group, m and n are the same or different and are each nought or one, and  $D_1$  and  $D_2$  are the same or different and are each a residue of a five-membered or six-membered heterocyclic nitrogen nucleus.

It is to be understood that the heterocyclic nuclei may be polynuclear, the additional rings being themselves isocyclic. Thus each may contain a fused benzene ring as, for example, in a quinoline nucleus. Phenyl groups and fused isocyclic groups present in the compounds may themselves carry substituent groupings, e.g. alkyl, aryl, alkoxy, hydroxy, amino or acylamino groups or halogen atoms (e.g. chlorine or bromine).

Examples of heterocyclic nuclei of which  $D_1$  and  $D_2$  may be the residue are thiazole, pyridine, pyrimidine, thiadiazole and their partially reduced derivatives (e.g. thiazoline) and the polynuclear derivatives of these such as benzothiazole, quinoline and benziminazole. Where the residues  $D_1$  and  $D_2$  include tertiary nitrogen atoms of the form —NR<sup>1</sup>—, R<sup>1</sup> may be hydrogen, alkyl, aryl or aralkyl.

The group R<sub>1</sub> may be an alkyl group containing up to 4 carbon atoms and the group R<sub>2</sub> may be an alkyl group containing up to 4 carbon atoms or a phenyl group.

According to the present invention there are provided quaternary cyclammonium salts of the said compounds of formula I, being compounds of the general formula II:—

where the symbols R<sub>1</sub>, R<sub>2</sub>, D<sub>1</sub>, D<sub>2</sub>, m and n have the meanings assigned to them above, R<sub>3</sub> is an alkyl or aralkyl group and X is an anion, e.g. toluene-p-sulphonate, halide (chloride, bromide or iodide), sulphate, methosulphate, perchlorate or sulphamate.

Preferably the said compounds are pre-

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pared by treating a compound of general formula I with a quaternising agent of the formula R<sub>2</sub>X, e.g. by refluxing with or without an inert solvent, if R<sub>2</sub>X is of suitable volatility, or by fusing the reactants at elevated temperature.

The resulting dyes are valuable dyestuffs for

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polymeric materials consisting essentially of polymers of acrylonitrile and copolymers of acrylonitrile with other monomeric materials. Generally the polymers will contain at least 85 per cent of acrylonitrile units and a particular commercial material of the type, essentially in the form of textile fabric, is sold under the Registered Trade Mark ORLON. Such polymers are difficult to dye by the use of conventional dyestuffs and the compounds of the present invention provide a new class of dye-stuff of exceptional value in the dyeing of such materials. The dyeing is preferably effected in a dilute acid medium or in a dispersion.

The dyes of formula II may be rendered of suitable solubility by treating with an aqueous solution of a salt MY where M is ammonium or an atom of an alkali metal (e.g. sodium or potassium) and Y is an anion, e.g. any of those referred to in connection with X whereby the radicle X is replaced by the radicle Y so that the solubility of the product is varied. Where R<sub>1</sub> in the compound of formula I is hydrogen it is possible by appropriately selecting the conditions to alkylate that hydrogen atom and at the same time convert the product to a quaternary salt of general formula II.

The following Examples will serve to illustrate the invention: -

EXAMPLE 1.

(1:3 - Dimethyl - 2 - benziminazole) (31 methyl - 21 - benzothiazole) - 8 - methyl- $8^{1}:9^{1}$  - diazatrimethincyanine toluene - p sulphonate.

1 - (3 - Methyl - 2 - benzothiazolinylidene-2-( $\alpha$  - 1 - methyl - 2 - benziminazolylethylidene) hydrazine (2.0 g.) was fused with methyl toluene - p - sulphonate (4.0 g.) at 150° for 2 hours. The mixture was cooled and lixivated with dry acetone to give a yellow solid which was filtered off and recrystallised from ethanol to give the pure dye as fine yellow needles, m.pt, 209-12°,

Example 2.

(1 - Methyl - 2 - pyridine) (3¹ - methyl - 2¹-benzothiazole) - 7 - methyl - 8¹:9¹ - diazatrimethincyanine toluene-p-sulphonate 1 - (3 - Methyl - 2 - benzothiazolinylidene)- $2-(\alpha-2-pyridylethylidene)$  hydrazine (2.0 g.) and 50 methyl toluene-p-sulphonate (4.0 ml.) were fused at 150-160° for 2 hours. The resulting solid was washed with dry benzene, filtered off and recrystallised from ethanol to give the pure

55 Example 3. (1 - Methyl - 4 - pyridine) (31 - methyl - 21benzothiazole) - 7 - methyl - 81:91 - diazatrimethincyanine toluene-p-sulphonate

dye as small yellow needles, m.pt. 203-5°.

1 - (3 - Methyl - 2 - benzothiazolinylidene)-2-(a-4-pyridylethylidene) hydrazine (2.0 g.) and methyl toluene-p-sulphonate (4.0 g.) were fused at 150—160° for 2 hours to give an orange solid. This solid was ground with dry

benzene and then recrystallised from ethanol to give the pure dye as orange needles, m.pt. 233°.

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Example 4.

(1 - Methyl - 2 - quinoline) (3<sup>1</sup> - ethyl - 2<sup>1</sup>-benzothiazole) - 9 - methyl - 8<sup>1</sup>: 9<sup>1</sup> - diazatrimethincyanine bromide

1 - (3 - Ethyl - 2 - benzothiazolinylidene)-2(a-2-quinolylethylidene) hydrazine (2.5 g.) and methyl toluene-p-sulphonate (5.0 g.) were fused at 150° for 4 hours. The resulting orange melt was dissolved in ethanol and the solution was added to 25% aqueous sodium bromide (120 ml.) to give an orange solid. This solid was filtered off, ground with benzene and recrystallised from ethanol to give the pure dye as dark red needles, m.pt. 229°.

EXAMPLE 5.

Bis (3-methyl-2-benzothiazole) - 8 - methyl-81: 91-diazatrimethincyanine bromide

1 - (3 - Methyl - 2 - benzothiazolinylidene)- $2(\alpha-2-benzothiazolyl-ethylidene)$ hydrazine (1.55 g.) and methyl toluene-p-sulphonate (3.1 g.) were fused at 155—160° for 2 hours. The resulting deep red solid was ground with dry ether and then boiled out with acetone to give the crude dye toluene-p-sulphonate. This salt was dissolved in methanol (20 ml.) and added to 15% sodium bromide (20 ml.) to give a red solid which was filtered off and recrystallised from water to give the pure dye as orange-red needles, m.pt. 207°.

The following dyes were prepared from the bases of the co-pending Application referred to above by fusing with methyl toluene-psulphonate in the manner of Example 1.

EXAMPLE 6.

(3 - Methyl - 2 - benzothiazole) (11 - methyl-41 - pyridine) - 71 - phenyl - 8:9 - diazatrimethincyanine toluene-p-sulphonate was obtained from  $1 - (3 - \text{methyl} - 2 - \text{benzothiazolinylidene}) - 2(\alpha - 4 - \text{pyridylbenzylidene})$ hydrazine and crystallised from ethanol as orange leaflets, m.pt. 214°.

Example 7.

(3 - Methyl - 2 - benzothiazole) (11 - methyl-2 - quinoline) - 91 - benzyl - 8:9 - diazatrimethincyanine toluene-p-sulphonate was obtained from 1 - (3 - methyl - 2 - benzothiazolinylidene) -  $2(\alpha - 2 - quinolyl - \beta - phenyl$ ethylidene) hydrazine and crystallised from ethanol as red microcrystals, m.pt. 249—52°. 115

EXAMPLE 8.

(6 - Ethoxy - 3 - methyl - 2 - benzothiazole) (1<sup>1</sup> - methyl - 4<sup>1</sup> - pyridine) - 7<sup>1</sup> - methyl-8:9 - diazatrimethincyanine toluene - p sulphonate was obtained from 1-(6-ethoxy- 120 3-methyl-2-benzothiazolinylidene –  $2(\alpha - 4$  - pyridylethylidene hydrazine and crystallised from ethanol as orange-red needles, m.pt.

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	T-V/MAT	LE 7.		
(1 - Methyl -	2 - quinol	ine) (31 –	methyl .	_ 2¹-
benzothiazo	le) - 8 <sup>1</sup> -	methyl -	9:10-di	iaza-
trimethincy	anine tolu	ene-p-sulp	honate	was
obtained from	1-(1:2-	dihydro	- 1 - me	thyl-
2 – quinolinyli				
ethylidene) h				
ethanol as ora				

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EXAMPLE 10.

10 (6 - Methoxy - 3 - methyl - 2 - benzothiazole)
 (3¹ - methyl - 2¹ - benzothiazole) - 8¹-methyl - 8:9 - diazatrimethincyanine toluene-p-sulphonate was obtained from 1-(6 - methoxy - 3 - methyl - 2 - benzothiazolinylidene) - 2 - (a - 2 - benzothiazolylethylidene) hydrazine and crystallised from ethanol as red plates, m.pt. 209—11°.

#### EXAMPLE 11.

(5:6 - Dimethoxy - 3 - methyl - 2 - benzothiazole)
 8¹ - methyl - 8:9 - diazamethincyaninetoluene-p-sulphonate was obtained from 1 (5:6 - dimethoxy - 3 - methyl - 2 - benzothizolinylidene - 2 - (α - 2 - benzothiazolylethylidene)
 bydrazine and crystallised from ethanolas red plates with a green reflex, m.pt. 219—222°.

EXAMPLE 12.

(3 - Methyl - 5 - methylthio - 2-1:3:4 - thiadiazole) (3¹ - methyl - 2¹ - benzothiazole)-8¹ - methyl - 6:7 - diazatrimethincyanine toluene-p-sulphonate was obtained from 1-(3 - methyl - 5 - methylthio - 2 - Δ⁴-1:3:4-thiadiazolinylidene) - 2(α - 2 - benzothiazolylethylidene) hydrazine and crystallised from ethanol as light orange plates, m.pt. 203—6°.

#### EXAMPLE 13.

(1 - Methyl - 2 - quinoline) (1<sup>1</sup> - ethyl - 4<sup>1</sup>pyridine) 7<sup>1</sup> - phenyl - 9: 10 - diazatrimethincyanine toluene-p-sulphonate.

1 - (1:2 - Dihydro - 1 - methyt - 2 - quinolinylidene) - 2 - (a - 4 - pyridylbenzylidene) hydrazine (1.0 g.) and ethyl toluene-p-sulphonate (2.0 g.) were fused together at 150° for 4 hours. Dry acetone (50 ml.) was added to the melt and the dye slowly crystallised out. The solid was filtered off and dried to give the product as small red plates, m.pt. 180—5°.

#### Example 14.

50 (3 - Methyl - 2 - benzothiazole) (3¹ - methyl-2¹ - benzothiazole) - 8¹ - ethyl - 8∶9 - diazatrimethincyanine iodide was obtained from 1-(3-methyl-2-benzothiazolinylidene)-2-(α - 2 - benzothiazolylethylidene) hydrazine by fusing with methyl toluene-p-sulphonate in the manner of Example 1, precipitated with

potassium iodide in methanol and crystallised from methanol as red needles, m.pt. 243-4°.

Example 15.

(1 - Methyl - 2 - quinoline) (1<sup>1</sup> - benzyl - 4<sup>1</sup> - 60 pyridine) - 7<sup>1</sup> - phenyl - 9:10 - diazatrimethincyanine bromide

1(1:2 - Dihydro - 1 - methyl - 2 - quinolinylidene) - 2(a - 4 - pyridylbenzylidene) hydrazine (1.0 g.), chloroform (10 ml.) and benzyl bromide (2 ml.) were boiled under reflux with a calcium chloride guard tube for 6 hours. Dilution with dry acetone slowly gave the product as orange-red plates, m.pt. 236— 237° after filtration and drying.

EXAMPLE 16.

(6 - Ethoxy - 3 - methyl - 2 - benzothiazole) (3¹ - methyl - 2¹ - benzothiazole) - 8¹methyl - 8:9 - diazatrimethincyanine methosulphate

1 - (6 - Ethoxy - 3 - methyl - 2 - benzothiazolinylidene)-2 - (a - 2 - benzothiazolylethylidene) hydrazine (1.0 g.) was dissolved in boiling dry benzene (20 ml.) and dimethyl sulphate (1.0 ml.) added. After boiling for 16 hours, a mass of dark crystals were present and were filtered off, washed well with benzene and purified by boiling out with acetone to give the pure dye as dark red needles, m.pt. 205—

EXAMPLE 17.

(3 - Methyl - 2 - benzothiazole) (1¹ - methyl-4¹ - pyridine) - 7¹ - phenyl - 8:9 - diazatrimethincyanine methylsulphate was obtained by a method similar to that of Example 16 from 1 - (3 - methyl - 2 - benzothiazolinylidene) - 2 - (α - 4 - pyridylbenzylidene) hydrazine and crystallised from ethanol as yellow needles, m.pt. 227°.

EXAMPLE 18.

(1 - Methyl - 4 - quinoline) (3¹ - methyl - 2¹-benzothiazole) - 8¹ - methyl - 9:10 - diazatrimethincyanine toluene-p-sulphonate was obtained from 1(1:4-dihydro-1-methyl-4-quinolinylidene) - 2 - (α - 2 - benzothiazolyl-ethylidene) hydrazine as bronze leaflets from water, m.pt. 215—8°.

Example 19.

(2 - Methyl - 6 - phenyl - 3 - pyridazine)
(3¹ - methyl - 2¹ - benzothiazole) - 8¹ - 105
methyl - 7:8 - diazatrimethincyanine
methosulphate was obtained from 1(2-methyl
2:3 - dihydro - 6 - phenyl - 3 - pyridazinylidene)2(α-2-benzothiazolylethylidene) hydrazine
by a method similar to that of Example 16 and
purified by boiling out with acetone to give
the pure dye as red microcrystals, m.pt. 218°.

EXAMPLE 20.

(5 - Chloro - 3 - methyl - 2 - benzothiazole) (3¹ - methyl - 2¹ - benzothiazole) - 8¹ - 115 methyl - 8:9 - diazatrimethincyanine metho-

sulphate was obtained from 1(5-chloro-3methyl - 2 - benzothiazolinylidene) -  $2(\alpha - 2 - benzothiazolylethylidene)$  hydrazine by a method similar to that of Example 16 and purified by boiling out with acetone to give the pure dye as red microneedles, m.pt. 264-

EXAMPLE 21. Bis (3 - Methyl - 2 - benzothiazole) - 81methyl - 8:9 - diazatrimethincyanine iodide 1 - ( $\hat{2}$  - Benzothiazolyl) - 2 - ( $\alpha$  - 2 - benzothiazolylethylidene) hydrazine (2.5 g.), methyl iodide (5 ml.) and methanol (10 ml.) were boiled under reflux for 40 hours. The mixture was cooled and the crystals filtered off. After washing with cold methanol the solid was extracted in a soxhlet apparatus with boiling benzene to remove the last traces of the original base. The solid remaining was recrystallised from methanol to give the pure dye as small orange leaflets, m.pt. 194-6°.

EXAMPLE 22.

(3 - Methyl - 2 - benzothiazole) (11 - methyl-21 - quinoline) - 91 - phenyl - 8:9 - diazatrimethincyanine toluene-p-sulphonate 1(3 - Methyl - 2 - benzothiazolinylidene)2- $(\alpha - 2 - quinolylbenzylidene)$  hydrazine (1.5 g.) and methyl toluene-p-sulphonate (3.0 g.) were fused in an oil bath for 1 hour at 130° The melt was dissolved in dry acetone and allowed to stand. An orange crystalline mass

Example 2, " Example 3 " " Example 22 " 23

The present invention accordingly includes the compounds of general formula II, their production by the methods described, the dye-ing of textile materials, particularly polyacrylonitrile materials, by means of said com-

was present the next day and was filtered off and washed well with acetone to give the pure dye as red microcrystals, m.pt. 118°.

EXAMPLE 23.

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This Example illustrates the dyeing of polyacrylonitrile fibres with the compounds of the

present invention: -0.5 Parts of the dyestuff obtained according to Example 5 are pasted with 0.5 parts of 80% acetic acid and dissolved by the addition of 4000 parts of hot water. A further part of 80% acetic acid, 2 parts of sodium acetate and 4 parts of a condensation product from olein alcohol and 15 mols of ethylene oxide are 45 added and 100 parts of polyacrylonitrile fibres are entered. The bath is heated to 90° within 30 minutes, kept at this temperature for 10 minutes and then dyeing is performed at the boil for 1 hour. The dyebath is almost completely exhausted. The dyed goods are then soaped for 15 minutes at 80° in 5000 parts of water with the addition of a sulphonated fatty acid condensation product, rinsed and dried. The polyacrylonitrile fibres are dyed in a pure orange shade which has excellent fastness to

washing and light. Preferred compounds for use in the dyeing of such fibres are those of Examples 1, 2, 3, 5 and 22, and, by using in a similar manner the dyes of these Examples, the following shades are obtained:-

Dyeings of compound of Example 1 on polyacrylonitrile are greenish-yellow.

" yellow. " orange-yellow.

" orange.

pounds and the dyed materials so obtained.

WHAT WE CLAIM IS:-1. Diazacarbocyanines of the general formula: -

wherein R<sub>1</sub> is an alkyl or aralkyl group, R<sub>2</sub> is an alkyl, aryl or aralkyl group, m and n are the same or different and are each nought or one, D1 and D2 are the same or different and are each a residue of a five-membered or sixmembered heterocyclic nitrogen nucleus, R, is an alkyl or aralkyl group, and X is an anion.

2. Diazacarbocyanines according to claim I wherein R1 is an alkyl group containing up to 4 carbon atoms.

3. Diazacarbocyanines according to claim 1 or 2 wherein R2 is an alkyl group containing

up to 4 carbon atoms, or a phenyl group.
4. (1:3 - Dimethyl - 2 - benziminazole) (31methyl - 21 - benzothiazole) - 8 - methyl81:91 - diazatrimethincyanine toluene - psulphonate.

5. (1 - Methyl - 2 - pyridine) (3<sup>1</sup> - methyl-2<sup>1</sup> - benzothiazoline) - 7 - methyl - 8<sup>1</sup>:9<sup>1</sup>diazatrimethincyanine toluene - p - sulphonate.

6. (1 - Methyl - 4 - pyridine) (3<sup>1</sup> - methyl-2<sup>1</sup> - benzothiazole) - 7 - methyl - 8<sup>1</sup>:9<sup>1</sup>-diaza-

trimethincyanine toluene-p-sulphonate.
7. Bis (3 - methyl - 2 - benzothiazole) - 8methyl-81:91-diazatrimethincyanine bromide.

8. (3 - Methyl - 2 - benzothiazole) (1<sup>1</sup>-methyl - 2<sup>1</sup> - quinoline) - 9<sup>1</sup> - phenyl - 8:9diazatrimethincyanine toluene-p-sulphonate.

9. A process for the production of a diazacarbocyanine as defined in claim 1 which com-

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prises reacting a compound of the general formula: -

where the symbols have the meanings assigned to them in claim 1, with a quaternising agent of the formula R,X where the symbols have the meanings assigned to them in claim 1.

10. A process according to claim 9 wherein the reaction is effected by refluxing the re-

actants together.

11. A process according to claim 9 wherein the reaction is effected by fusing the reactants together,

12. A process for dyeing polymeric materials comprising at least 85% of acrylonitrile units which comprises subjecting such materials to treatment with a solution in a dilute acid medium, or with a dispersion, of a diazacarbocyanine as defined in any of Claims 1-8.

13. Polymeric materials comprising at least 85% of acrylonitrile units dyed with a diazacarbocyanine as defined in any of claims 1-8.

14. A diazacarbocyanine according to claim 1 as specifically set forth in any one of the foregoing specific Examples 4 and 6-21 inclusive.

15. A process for the production of a diazacarbocyanine as defined in claim 1 substantially as set forth in any one of the foregoing specific Examples 1—22.

16. A process according to claim 12 substantially as set forth in the foregoing specific Example 23.

> V. GALLAFENT, Chartered Patent Agent, 23, Roden Street, Ilford, Essex.

## PROVISIONAL SPECIFICATION

### **Diazacarbocyanines**

We, J. R. GEIGY S.A., a Swiss Company of Basle, Switzerland, do hereby declare this in-35 vention to be described in the following state-

This invention relates to new diazacarbo-

cyanine dyes.

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In our co-pending Application No. (Case A) there is described, inter alia, a new class of diazacarbocyanine bases of the general formula

wherein R<sub>1</sub> is selected from alkyl, aryl, aralkyl, hydroxyalkyl and carboxyalkyl groups,  $R_2$  is an alkyl, aryl or aralkyl group,  $Z_1$  and  $Z_2$  are the same or different and are each a methine (—CH=) group or a nitrogen atom, m and n

are the same or different and are each nought or one, and D1 and D2 are the same or different and are each a residue of a five-membered or six-membered heterocyclic nitrogen nucleus. It is to be understood that the heterocyclic nuclei may be polynuclear, the additional rings

being themselves heterocyclic or isocyclic. Thus each may contain a fused benzene ring as, for example, in a quinoline nucleus. Phenyl groups and fused isocyclic or heterocyclic groups present in the compounds may themselves carry substituent groupings, e.g. alkyl, aryl, alkoxy, hydroxy, amino or acylamino groups or halogen atoms (e.g. chlorine or bromine).

Examples of heterocyclic nuclei of which D<sub>1</sub> and D2 may be the residue are oxazole, thiazole, selenazole, pyridine, pyrrolenine, glyoxaline, pyrimidine, pyrazole, thiadiazole, triazole and their partially reduced derivatives (e.g. thiazoline) and the polynuclear derivatives of these such as benzothiazole, quinoline, indolenine, benziminazole and quinazoline. Where the residues D<sub>1</sub> and D<sub>2</sub> include tertiary nitrogen atoms of the form —NR<sup>1</sup>—, R<sup>1</sup> may be hydrogen, alkyl, aryl or aralkyl.

The group R<sub>1</sub> may be an alkyl or hydroxyalkyl group containing up to 4 carbon atoms and the group R2 may be an alkyl group containing up to 4 carbon atoms or a phenyl group.

According to the present invention there are provided quaternary ammonium salts of the said compounds of formula I, being compounds of the general formula II: -

$$\begin{bmatrix} ----D_1 - --- \\ N-(CH=Z_I)_m - C = N-N=C-C=(Z_2-CH)_n = N \\ R_1 & R_3 \end{bmatrix}^+ \times^-$$

.....II

where the symbols R<sub>1</sub>, R<sub>2</sub>, D<sub>1</sub>, D<sub>2</sub>, Z<sub>1</sub>, Z<sub>2</sub>, m and n have the meanings assigned to them above, R, is an alkyl, aralkyl, hydroxyalkyl, carboxyalkyl or aryl group and X is the amon of an acid, e.g. toluene-p-sulphonate, halide (chloride, bromide or iodide), sulphate, methosulphate, perchlorate or sulphamate.

Preferably the said compounds are prepared by treating a compound of general formula I with a quaternising agent of the formula R<sub>3</sub>X, e.g. by refluxing with or without an inert solvent, if R<sub>2</sub>X is of suitable volatility, or by fusing the reactants at elevated temperature.

The resulting dyes are valuable dyestuffs for polymeric materials consisting essentially of polymers of acrylonitrile and copolymers of acrylonitrile with other monomeric materials. Generally the polymers will contain at least 85 per cent of acrylonitrile units and a particular commercial material of the type, essentially in the form of textile fabric, is sold under the trade name ORLON. Such polymers are difficult to dye by the use of conventional dyestuffs and the compounds of the present invention provide a new class of dyestuff of exceptional value in the dyeing of such materials. The dyeing is preferably effected in a dilute acid medium.

The dyes of formula II may be rendered of suitable solubility by treating with an aqueous solution of a salt MY where M is ammonium or an atom of an alkali metal (e.g. sodium or potassium) and Y is an anion, e.g. any of those referred to in connection with X whereby the radicle X is replaced by the radicle Y so that the solubility of the product is varied. Where R<sub>1</sub> in the compound of formula I is hydrogen it is possible by appropriately selecting the conditions to alkylate that hydrogen atom and at the same time convert the product to a quaternary salt of general formula II.

The following Examples will serve to illus-

trate the invention:

EXAMPLE 1. (1:3 - Dimethyl - 2 - benziminazole) (31methyl - 21 - benzothiazole) - 8 - methyl- $2^{1}:9^{1}$  - diazatrimethincyanine toluene - p-

sulphonate.

1 - (3 - Methyl - 2 - benzothiazolinylidene)-2-(x-1-methyl - 2 - benziminazolylethylidene) hydrazine (2.0 g.) (Example 2 of our copending application) was fused with methyl toluene-psulphonate (4.0 g.) at 150° for 2 hours. The mixture was cooled and lixiviated with dry acetone to give a yellow solid which was filtered off and recrystallised from ethanol to give the pure dye as fine yellow needles, m.pt. 209ī2°.

EXAMPLE 2.

(1 - Methyl - 2 - pyridyl) (31 - methyl - 21benzothiazolyl) - 7 - methyl - 81:91-diaza-

trimethincyanine toluene-p-sulphonate. 1 - (3 - Methyl - 2 - benzothiazolinylidene)-2-(x-2-pyridyl-ethylidene) hydrazine (2.0 g.) (Example 3 of our co-pending application) and methyl toluene-p-sulphonate (4.0 ml.) was fused at 150-160° for 2 hours. The resulting solid was washed with dry benzene, filtered off and recrystallised from ethanol to give the pure dye as small yellow needles, m.pt. 203-5°.

EXAMPLE 3. (1 - Methyl - 4 - pyridyl) (3<sup>1</sup> - methyl - 2<sup>1</sup>-benzothiazole) - 7 - methyl - 8<sup>1</sup>: 9<sup>1</sup> - diazatrimethincyanine toluene-p-sulphonate.

1 - (3 - Methyl - 2 - benzothiazolinylidene)-2-(a-4-pyridylethylidene) hydrazine (2.0 g.) (Example 4 of our co-pending application) and methyl toluene-p-sulphonate (4.0 g.) were fused at 150-160° for 2 hours to give an orange solid. This solid was ground with dry benzene and then recrystallised from ethanol to give the pure dye as orange needles, m.pt. 233°. EXAMPLE 4.

(1 - Methyl - 2 - quinoline) (31 - ethyl - 21benzothiazole) - 9 - methyl - 81: 91 - diaza-

trimethincyanine bromide.

1 - (3 - Ethyl - 2 - benzothiazolinylidene)-2(α - 2 - quinolylethylidene) hydrazine (Example 5 of our co-pending application) (2.5 g.) and methyl toluene-p-sulphonate (5.0 g.) were fused at 150° for 4 hours. The resulting orange melt was dissolved in ethanol and the solution was added to 25% aqueous sodium bromide (120 ml.) to give an orange solid. This solid was filtered off, ground with benzene and recrystallised from ethanol to give the pure dye as dark red needles, m.pt. 229°. EXAMPLE 5.

Bis (3 - methyl - 2 - benzothiazole) - 8-methyl-81: 91-diazatrimethincyanine bromide.

1 - (3 - Methyl - 2 - benzothiazolinylidene)-2(x-2-benzothiazolylethylidene) hydrazine (1.55 g.) (Example 9 of our copending application) and methyl toluene-p-sulphonate (3.1 g.) were fused at 155—160° for 2 hours. The resulting deep red solid was ground with dry ether and then boiled out with acetone to give the crude dye toluene-p-sulphonate. This salt was dissolved in methanol (20 ml.) and added to 15% sodium bromide (20 ml.) to give a red solid which was filtered off and recrystallised from water to give the pure dye as orangered needles, m.pt. 207°.

The present invention accordingly includes the compounds of general formula II, their production by the method described, the dyeing of textile materials, particularly polyacrylonitrile materials, by means of said compounds and the dyed materials so obtained.

V. GALLAFENT, Chartered Patent Agent, 23, Roden Street, Ilford, Essex.

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